

Kimyo fanidan online test

1. (16). Tarkibida 30% CO va 70% N₂ bo'lgan gaz aralashmasini zichligini toping(g/ml).

1,25

1,96

1,52

0,98

2. (17). Quydagi qaysi tuzlar gidrolizlansa muhit kislotali bo'ladi? 1) kaliy permanganat, 2) seziiy ftorid, 3) ammoniy yodid, 4) natriy perxlorat, 5) aluminiy sulfit, 6) natriy sulfat, 7) kaliy xlorid, 8) ammoniy xlorid, 9) natriy nitrat, 10) temir (II) nitrat.

2,4,7

3,8,10

3,5,6

1,5,6

3. (18). Havodagi kislorodning hajmiy ulushini belgilang?

32 %

21 %

18 %

41 %

4. (19). Kislotali oksidlar qatorini ko`rsating.

Na₂O, MgO, BaO

ZnO, BeO, Al₂O₃

Mn₂O₇, P₂O₅, CO₂

K₂O, SO₃, Al₂O₃

5. (20). Quyidagi qaysi elektrolit dissotsiyalanganda eng ko'p ionlarga parchalanadi?

natriy gidroksidi

temir (III) sulfat

karbonat kislota

rux xloridi

6. (21). Ohaktosh CaCO_3 tarkibidagi uglerodning massa ulushini aniqlang.

40%

30%

12%

18%

7. (22). 350 ml (n.sh) da fosfor bug'lari 0,97 gr kelsa bug'dagi fosfor molekularining tarkibi qanday bo'ladi?

P_2

P_6

P_4

P

8. (23). Teng hajmli argon va azotni kislorodga nisbatan zichligini toping.

1,34

1,68

1,06

2,12

9. (24). 1 tonna sulfat kislota ishlab chiqarish uchun necha kg pirit kerak?

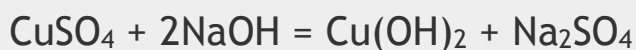
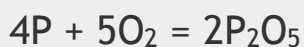
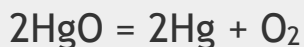
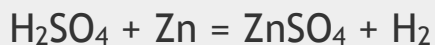
918 kg

612 kg

306 kg

1224 kg

10. (25). O'rin olish reaksiyasini aniqlang?



11. (26). Energetik pog'onadagi elektronlar bo'lishi mumkin bo'lgan maksimal yacheykalar soni qaysi formula bilan aniqlangan?

$2n^2$

n^2

$2l+1$

$2(2l+1)$

12. (27). Yil davomida odam 7 kg osh tuzi iste'mol

qilganda, har kuni qabul qiladigan Cl^- ionlari massasini hisoblang.

9,8

11,6

8,6

12,2

13. (28). Erkin vodorod gazi qayerda eng ko'p tarqalgan?

suvda

havoda

koinotda

yerda

14. (29). Elementlarning qaysi qatori atom radiusi ortishi tartibida keltirilgan?

C, N, O, F

O, S, Se, Te

Na, Mg, Al, Si

J, Br, Cl, F

15. (30). 5,6 litr SO_2 gazini n.sh da massasini to'ing.

32

16

64

6

